

# Contra-CEST

## Sunrise Debate: Clinical Utility of CEST in Oncology Course

**Speaker Name:** Kim Desmond, [kdesmond@sri.utoronto.ca](mailto:kdesmond@sri.utoronto.ca)

**Target audience:** Those with a basic understanding of the method who are considering implementing a CEST imaging protocol from an oncological perspective.

**Objective:** To acknowledge the negative aspects of CEST imaging and pitfalls that may be encountered as an aid to deciding whether CEST is a worthwhile approach to explore for the diagnosis of cancer and the assessment of therapy.

### Introduction

With all the buzz surrounding the topic of chemical exchange saturation transfer (CEST) as a “new”, molecule-specific contrast mechanism with promises of pH imaging, it is no surprise that there are many people interested in implementing it. This talk will provide a reality check, discussing the caveats that go along with the expectations, and the many challenges that stand in the way of an ideal CEST protocol.

CEST contrast can be separated into two broad categories: 1. endogenous CEST[1, 2] and CEST from biological molecules [3, 4], and 2. exogenous CEST contrast agents[5, 6]. Endogenous CEST arises from molecules native to the cells: amide protons on the protein backbone, glycosaminoglycans in collagen, etc. These protons resonate between approx. 0-5 ppm of the water resonance. Exogenous CEST contrast agents such PARACEST are typically chelates of paramagnetic ions (often a lanthanide), which can shift the resonance frequency into the hundreds of ppm. Each group shares common concerns related to nomenclature, sensitivity and specificity, protocol development, RF power, and data analysis, each from a slightly different perspective due to differences in concentration, resonance frequency and exchange rate of the labile protons. The exogenous contrast agents have additional issues regarding biocompatibility.

### Sensitivity and specificity

The sensitivity of CEST can be defined as CNR/unit concentration (a.k.a. proton transfer enhancement[4]) of the detectable moiety, with further refinements taking scan time into consideration (CNR efficiency[7]). In an ideal case, the bulk water protons could be completely and exclusively saturated through transfer of saturation from the CEST protons[8]. This is physically impossible due to finite relaxation times, resulting in saturation efficiency below 100%, and there are further limitations due to low concentration. PARACEST applications anticipate CEST-specific signal changes of upwards of 50% from phantom studies [5, 9, 10], but this is hardly realized in preclinical applications in which signal changes in tumours do not exceed 5-10%[11, 12]. Endogenous CEST is limited to approximately 5% in tumours[9]. Specificity and sensitivity are hampered by the presence of a multitude of additional factors which affect the contrast, including RF power[13], T1 and T2 relaxation, field inhomogeneities[14], other CEST pools[15, 16], and magnetization transfer from semisolid protons[17]. Several methods have been developed which are less sensitive to instrumental and other non-CEST related parameters[18-21], however the most failsafe method for removal of these terms involves the time-consuming acquisition of parameter maps, requiring the addition of  $T_1$ ,  $B_1$ ,  $B_0$  etc. mapping to the protocol[17]. To make matters worse, when employing the most straightforward CEST quantification (CESTR asymm) the endogenous CEST from amide protons which resonate at 3.5 ppm is influenced by the presence of NOE

from aliphatic protons which resonate at  $\sim$ -3.5 ppm[22] which is also increased in tumours. Endogenous CEST is specific to a type of labile group (amide, amine, hydroxyl, etc) however there are many subclassifications so that all the endogenous CEST spectra are composed of groups of compound peaks[23], limiting the ultimate specificity of the technique. Since labile proton relative concentration or exchange rate are correlated in the CEST model[25, 26], it may be impossible to discern which property is causing the observed contrast changes, especially when both protein concentration and pH are expected to change within the tumour. Furthermore, with endogenous CEST, the low CNR severely limits the accuracy and precision of fitting to quantitative models. The majority of CEST experiments involve the generation of “negative contrast”, in that the saturated signal is less than the reference signal. This poses a problem for CEST interpretation, and post-processing must occur so that regions of high CEST “light up” in an image instead of appearing darker. It also has the consequence that images with more CEST will have a lower SNR due to the reduced signal.

### **RF power**

In order to obtain the maximum saturation efficiency[27, 28], it may require a saturation pulse with an RF amplitude which exceeds the safety limits [29-31]. RF amplitudes ranging from 1 – 20  $\mu$ T have been used in preclinical studies of CEST in cancer. Even if a strong RF pulse was permissible, spillover effects will cause the observed CEST effect to be reduced by direct saturation of the water resonance[32]. There may not be an acceptable compromise between RF power limitations and minimum acceptable CNR. Increasing the field strength will increase the separation (in Hz) between CEST peaks and the water resonance, and provide increased SNR, however the deposited RF energy increases approximately with the square of the RF amplitude causing further restrictions on the CEST experiment.

### **Protocol and Analysis**

As of this writing, CEST protocols are not available by default with any of the vendor’s standard pulse sequence packages, and consequently a considerable amount of expertise is required to set it up at a new location. Often, CEST pulse sequence development involves overcoming limitations on maximum pulse length, and before the protocol becomes reasonably “user friendly” scripts or automation must be implemented so that the manifold images required for a CEST spectrum can be grouped and the offset frequencies easily assigned. Protocols optimized for 3 T [27, 33, 34] must be completely revised for 7 T[35] in light of the tighter restrictions on RF power. Since most vendors specify RF frequency in terms of Hz, rather than ppm, offset frequencies must also be recalculated when the field strength changes.

Despite this fact, numerous institutions have implemented the CEST sequence, and preclinical[15, 36-39] and clinical trials[27, 40-46] are underway. As CEST applications are in their infancy, with each new implementation improvements are made to the protocol and as a result there is no standard method. Optimizable parameters include the RF pulse strength and shape, TR, and choice of offset frequencies [10, 47, 48]. This limits the ability to compare CEST results between studies, especially in cases where the metric is “CEST contrast” (i.e. CESTR asymm[49] or peak amplitude[15, 21]) rather than a fundamental quantifiable property such as CEST pool concentration or exchange rate constants[50, 51]. Efforts have been made to reduce the scan time for CEST protocols[34, 52-54], but they largely remain time consuming and difficult to fit into a reasonable (i.e. 20 minutes) scan period, much less have sufficient temporal resolution[55] to capture immediate dynamic changes due to treatment.

Depending upon the completeness of the acquired CEST spectrum data, there are several options for correction of artifacts and inhomogeneities[14, 56-58], and a choice of CEST metrics to be determined. Full parameter quantification, however, of a multi-compartment Bloch equation model is impractical (difficult to implement, time consuming to run fitting). Parameters may be fixed to save processing time,

however deciding on what parameters to fix at what value and the impact of this choice upon the unfixed parameters is not immediately clear.

### **Nomenclature**

CEST Z-Spectra are preferentially displayed in order of decreasing offset frequency from left to right. This has its origins in NMR spectroscopy, where historically the axis was ordered from low shielding to high and the offset frequency is highest for the lowest amount of shielding. Ordering the x-axis from positive to negative values is the source of much confusion for those approaching CEST from a non-spectroscopic background. Worse, if one was to mistakenly switch the axis orientation, the presence of the aliphatic NOE at the incorrectly assumed position of amide CEST is misleading to the extent that the error may not be immediately realized. Even for those who are familiar with the axis convention, the reference point of the ppm axis is changed for CEST so that instead of a reference chemical (such as TMS) being denoted as 0, this is now shifted so that water protons are at 0 ppm.

There are many different quantitative and semi-quantitative methods of expressing the CEST metric, and one must take care to recognize the differences between CEST asymmetry[32], relative CEST pool size[59], and CEST peak amplitude[60] (to list a few). There are two schools of thought for expression of the exchange rate, as forward and backwards rates  $k_{sw}/k_{ws}$  [59] or to wrap these into a single pseudo-first order rate constant,  $R$ [61]. There are multitudes of (not necessarily mutually exclusive) applications with the naming convention “\_\_CEST”, referring to the labile exchange group or another distinguishing quality of the CEST contrast agent [62-65]. These imaging applications all follow similar protocols, but have their own set of optimized scanner parameters.

### **Biology and Chemistry**

CEST protocol development and experimentation requires a solid background in biochemistry (and chemical synthesis for the contrast agents), in addition to MRI physics. Knowledge of labile species, range of possible exchange rates and how these vary with solvent accessibility and pH are a few of the major issues. CEST contrast agents must be engineered to have exchange rates within an optimal window and with the offset frequency as large as possible[50, 66]. It is difficult to replicate the biological conditions and thus CEST contrast agents tested in phantoms often have markedly better performance than in vivo where unpredictable bonding may occur[67] and there is the presence of competing magnetization transfer pathways[9].

In order to interpret CEST images of cancer, one must be aware of which biological compartment is measured by the CEST experiment – mainly intracellular for endogenous amide CEST due to the higher concentration of proteins and peptides[2], and extracellular for PARACEST due to the accumulation of extravasated contrast without uptake by the cells[68-70]. Intracellular and extracellular pH behave differently depending on the pathology or treatment [71-75], and the sensitivity of the CEST experiment to changes in these compartments must be determined prior to embarking on large scale studies. Changes in pH (especially intracellular) are small due to the mechanisms for maintenance of homeostasis[76].

### **CEST contrast agents**

An immediate concern with CEST contrast agents (namely PARACEST), as with any injected compound, is toxicity to humans and the associated lengthy approval process for those deemed safe in pilot preclinical trials. PARACEST agents are similar in structure to Gd-containing counterparts used as T1-shortening agents; the chemistry is also similar and the stability of the compounds is more closely related to the ligand/chelate than the lanthanide ion[67]. Paramagnetic ions are coated, chelated or

incorporated into macromolecular structures during preparation, which all but eliminates their toxicity[77]. However, PARACEST agents larger than 5nm evade rapid clearance by the kidneys[78] and high concentrations can accumulate in the tissue once leaving the blood pool[79]. Over time, if the agents remain in the system (often in patients with impaired kidney function) the agents can break down to reveal the toxic core.

### **Oncological applications of CEST and alternatives**

CEST has been proposed as an MRI contrast mechanism sensitive to protein concentration and pH, however the question remains whether it is more effective than pre-existing imaging methods with similar objectives. Although CEST performs well at identifying regions of tumour, with the exception of recurrent cancer in the presence of radiation necrosis[80], it does not appear that CEST offers improved diagnostic potential over standard methods (i.e. T1- and T2- weighted imaging, DCE-MRI). There is potential in CEST for identification of hypoxic regions which have increased glycolysis[81] and may be resistant to treatment, but it is unclear whether this offers sufficient advantages over FDG-PET. Endogenous pH imaging is an enticing target for CEST experiments, however absolute pH imaging with CEST remains elusive[82, 83]. CEST images which are simply “pH-sensitive” may not be accurate enough to be helpful for evaluation of pH-triggered drugs[84-86].

### **References**

1. Zhou, J., et al., *Amide proton transfer (APT) contrast for imaging of brain tumors*. Magnetic Resonance in Medicine, 2003. **50**(6): p. 1120-1126.
2. Zhou, J., et al., *Using the amide proton signals of intracellular proteins and peptides to detect pH effects in MRI*. Nature medicine, 2003. **9**(8): p. 1085-1090.
3. Ward, K.M., A.H. Aletras, and R.S. Balaban, *A new class of contrast agents for MRI based on proton chemical exchange dependent saturation transfer (CEST)*. Journal of Magnetic Resonance, 2000. **143**(1): p. 79-87.
4. McMahon, M.T., et al., *New “multicolor” polypeptide diamagnetic chemical exchange saturation transfer (DIACEST) contrast agents for MRI*. Magnetic Resonance in Medicine, 2008. **60**(4): p. 803-812.
5. Woods, M., D.E. Woessner, and A.D. Sherry, *Paramagnetic lanthanide complexes as PARACEST agents for medical imaging*. Chemical Society Reviews, 2006. **35**(6): p. 500-11.
6. Aime, S., M. Fasano, and E. Terreno, *Lanthanide (III) chelates for NMR biomedical applications*. Chemical Society Reviews, 1998. **27**(1): p. 19-29.
7. Liu, G., et al., *PARACEST MRI with improved temporal resolution*. Magnetic Resonance in Medicine, 2009. **61**(2): p. 399-408.
8. Forsén, S. and R.A. Hoffman, *Study of moderately rapid chemical exchange reactions by means of nuclear magnetic double resonance*. The Journal of Chemical Physics, 1963. **39**: p. 2892-2901.
9. Li, A.X., et al., *Four-pool modeling of proton exchange processes in biological systems in the presence of MRI-paramagnetic chemical exchange saturation transfer (PARACEST) agents*. Magnetic Resonance in Medicine, 2008. **60**(5): p. 1197-1206.
10. Woessner, D.E., et al., *Numerical solution of the Bloch equations provides insights into the optimum design of PARACEST agents for MRI*. Magnetic Resonance in Medicine, 2005. **53**(4): p. 790–799.
11. Li, A.X., et al., *In vivo detection of MRI-PARACEST agents in mouse brain tumors at 9.4 T*. Magnetic Resonance in Medicine, 2011. **66**(1): p. 67-72.

12. Jones, C.K., et al., *In vivo detection of PARACEST agents with relaxation correction*. Magnetic Resonance in Medicine, 2010. **63**(5): p. 1184-1192.
13. Sun, P.Z., P.C. van Zijl, and J. Zhou, *Optimization of the irradiation power in chemical exchange dependent saturation transfer experiments*. Journal of Magnetic Resonance, 2005. **175**(2): p. 193-200.
14. Sun, P.Z., C.T. Farrar, and A.G. Sorensen, *Correction for artifacts induced by B0 and B1 field inhomogeneities in pH-sensitive chemical exchange saturation transfer (CEST) imaging*. Magnetic Resonance in Medicine, 2007. **58**(6): p. 1207-1215.
15. Desmond, K.L., F. Moosvi, and G.J. Stanisz, *Mapping of amide, amine, and aliphatic peaks in the CEST spectra of murine xenografts at 7 T*. Magnetic Resonance in Medicine, 2013.
16. Jones, C., A. Huang, and P. van Zijl. *Exchange-relayed nuclear Overhauser effect MRI*. in *Proceedings of 19th Annual Meeting ISMRM, Montreal, Canada*. 2011.
17. Desmond, K.L. and G.J. Stanisz, *Understanding quantitative pulsed CEST in the presence of MT*. Magnetic Resonance in Medicine, 2012. **67**(4): p. 979-90.
18. Zaiss, M., S. Goerke, and P. Bachert, *MT and spillover correction for quantitative steady-state pulsed CEST-MRI at 3T using the reciprocal Z-spectrum*. arXiv preprint arXiv:1302.6605, 2013.
19. Zu, Z., et al., *Imaging amide proton transfer and nuclear overhauser enhancement using chemical exchange rotation transfer (CERT)*. Magnetic Resonance in Medicine, 2013.
20. Scheidegger, R., E. Vinogradov, and D.C. Alsop, *Amide proton transfer imaging with improved robustness to magnetic field inhomogeneity and magnetization transfer asymmetry using saturation with frequency alternating RF irradiation*. Magnetic Resonance in Medicine, 2011. **66**(5): p. 1275-1285.
21. Zaiß, M., B. Schmitt, and P. Bachert, *Quantitative separation of CEST effect from magnetization transfer and spillover effects by Lorentzian-line-fit analysis of z-spectra*. Journal of Magnetic Resonance, 2011. **211**(2): p. 149-155.
22. Jin, T., et al., *MR imaging of the amide-proton transfer effect and the pH-insensitive nuclear overhauser effect at 9.4 T*. Magnetic Resonance in Medicine, 2012.
23. Denisov, V.P. and B. Halle, *Hydrogen exchange rates in proteins from water 1H transverse magnetic relaxation*. Journal of the American Chemical Society, 2002. **124**(35): p. 10264-10265.
24. *Statistics Calculated for All Chemical Shifts from Atoms in the 20 Common Amino Acids*. Biological Magnetic Resonance Data Bank [cited 2012; Available from: [http://www.bmrb.wisc.edu/ref\\_info/statful.htm](http://www.bmrb.wisc.edu/ref_info/statful.htm)].
25. Wu, R., et al., *Improved measurement of labile proton concentration-weighted chemical exchange rate (kws) with experimental factor-compensated and T1-normalized quantitative chemical exchange saturation transfer (CEST) MRI*. Contrast Media & Molecular Imaging, 2012. **7**(4): p. 384-389.
26. Dixon, W.T., et al., *A concentration-independent method to measure exchange rates in PARACEST agents*. Magnetic Resonance in Medicine, 2010. **63**(3): p. 625-632.
27. Sun, P.Z., et al., *Investigation of optimizing and translating pH-sensitive pulsed-chemical exchange saturation transfer (CEST) imaging to a 3T clinical scanner*. Magnetic Resonance in Medicine, 2008. **60**(4): p. 834-841.
28. Schmitt, B., et al., *Optimization of pulse train presaturation for CEST imaging in clinical scanners*. Magnetic Resonance in Medicine, 2011. **65**(6): p. 1620-1629.
29. Wang, Z., et al., *SAR and temperature: simulations and comparison to regulatory limits for MRI*. Journal of Magnetic Resonance Imaging, 2007. **26**(2): p. 437-441.
30. Collins, C.M., et al., *Temperature and SAR calculations for a human head within volume and surface coils at 64 and 300 MHz*. Journal of Magnetic Resonance Imaging, 2004. **19**(5): p. 650-656.

31. *Guidance for Industry and FDA Staff - Criteria for Significant Risk Investigations of Magnetic Resonance Diagnostic Devices* 2003 December 29, 2013]; Available from: <http://www.fda.gov/downloads/MedicalDevices/DeviceRegulationandGuidance/GuidanceDocuments/ucm072688.pdf>.
32. Sun, P.Z. and A.G. Sorensen, *Imaging pH using the chemical exchange saturation transfer (CEST) MRI: correction of concomitant RF irradiation effects to quantify CEST MRI for chemical exchange rate and pH*. *Magnetic Resonance in Medicine*, 2008. **60**(2): p. 390-397.
33. Dula, A.N., et al., *Amide proton transfer imaging of the breast at 3 T: Establishing reproducibility and possible feasibility assessing chemotherapy response*. *Magnetic Resonance in Medicine*, 2012.
34. Zhu, H., et al., *Fast 3D chemical exchange saturation transfer (CEST) imaging of the human brain*. *Magnetic Resonance in Medicine*, 2010. **64**(3): p. 638-644.
35. Klomp, D.W., et al., *Amide proton transfer imaging of the human breast at 7T: development and reproducibility*. *NMR in Biomedicine*, 2013.
36. Sheth, V.R., et al., *Measuring in vivo tumor pHe with CEST-FISP MRI*. *Magnetic Resonance in Medicine*, 2012. **67**(3): p. 760-768.
37. Wolf, G. and N. Abolmaali, *Preclinical Molecular Imaging Using PET and MRI*, in *Molecular Imaging in Oncology* 2013, Springer. p. 257-310.
38. Flament, J., et al., *In vivo CEST MR imaging of U87 mice brain tumor angiogenesis using targeted LipoCEST contrast agent at 7 T*. *Magnetic Resonance in Medicine*, 2013. **69**(1): p. 179-187.
39. Ferrauto, G., et al., *In vivo maps of extracellular pH in murine melanoma by CEST-MRI*. *Magnetic Resonance in Medicine*, 2013.
40. Ling, W., et al., *Assessment of glycosaminoglycan concentration in vivo by chemical exchange-dependent saturation transfer (gagCEST)*. *Proceedings of the National Academy of Sciences*, 2008. **105**(7): p. 2266-2270.
41. Mougin, O., et al., *Magnetization transfer phenomenon in the human brain at 7 T*. *NeuroImage*, 2010. **49**(1): p. 272-281.
42. Singh, A., et al., *Chemical exchange saturation transfer magnetic resonance imaging of human knee cartilage at 3 T and 7 T*. *Magnetic Resonance in Medicine*, 2012. **68**(2): p. 588-594.
43. Jia, G., et al., *Amide proton transfer MR imaging of prostate cancer: a preliminary study*. *Journal of Magnetic Resonance Imaging*, 2011. **33**(3): p. 647-654.
44. Zhou, J., et al., *Practical data acquisition method for human brain tumor amide proton transfer (APT) imaging*. *Magnetic Resonance in Medicine*, 2008. **60**(4): p. 842-849.
45. Haris, M., et al., *In vivo mapping of brain myo-inositol*. *NeuroImage*, 2011. **54**(3): p. 2079-2085.
46. Ren, J., et al., *Imaging the tissue distribution of glucose in livers using a PARACEST sensor*. *Magnetic Resonance in Medicine*, 2008. **60**(5): p. 1047-1055.
47. Zu, Z., et al., *Optimizing pulsed-chemical exchange saturation transfer imaging sequences*. *Magnetic Resonance in Medicine*, 2011. **66**(4): p. 1100-1108.
48. Tee, Y., et al., *Optimal sampling schedule for chemical exchange saturation transfer*. *Magnetic Resonance in Medicine*, 2013.
49. Zhou, J., et al., *Using the amide proton signals of intracellular proteins and peptides to detect pH effects in MRI*. *Nature Medicine*, 2003. **9**(8): p. 1085-90.
50. Woessner, D.E., et al., *Numerical solution of the Bloch equations provides insights into the optimum design of PARACEST agents for MRI*. *Magnetic Resonance in Medicine*, 2005. **53**(4): p. 790-799.

51. Sun, P.Z., *Simplified and scalable numerical solution for describing multi-pool chemical exchange saturation transfer (CEST) MRI contrast*. Journal of Magnetic Resonance, 2010. **205**(2): p. 235-241.
52. Sun, P.Z., et al., *Fast multislice pH-weighted chemical exchange saturation transfer (CEST) MRI with Unevenly segmented RF irradiation*. Magnetic Resonance in Medicine, 2011. **65**(2): p. 588-594.
53. Dixon, W.T., et al., *A multislice gradient echo pulse sequence for CEST imaging*. Magnetic Resonance in Medicine, 2010. **63**(1): p. 253-256.
54. Jones, C.K., et al., *In vivo three-dimensional whole-brain pulsed steady-state chemical exchange saturation transfer at 7 T*. Magnetic Resonance in Medicine, 2012. **67**(6): p. 1579-1589.
55. Shapiro, M.G., et al., *Dynamic imaging with MRI contrast agents: quantitative considerations*. Magnetic Resonance Imaging, 2006. **24**(4): p. 449-462.
56. Stancanello, J., et al., *Development and validation of a smoothing-splines-based correction method for improving the analysis of CEST-MR images*. Contrast Media & Molecular Imaging, 2008. **3**(4): p. 136-149.
57. Kim, M., et al., *Water saturation shift referencing (WASSR) for chemical exchange saturation transfer (CEST) experiments*. Magnetic Resonance in Medicine, 2009. **61**(6): p. 1441-1450.
58. Singh, A., et al., *On B1 inhomogeneity correction of in vivo human brain glutamate chemical exchange saturation transfer contrast at 7T*. Magnetic Resonance in Medicine, 2012.
59. Zhou, J. and P.C.M. van Zijl, *Chemical exchange saturation transfer imaging and spectroscopy*. Progress in Nuclear Magnetic Resonance Spectroscopy, 2006. **48**(2-3): p. 109-136.
60. Zaiss, M., B. Schmitt, and P. Bachert, *Quantitative separation of CEST effect from magnetization transfer and spillover effects by Lorentzian-line-fit analysis of z-spectra*. Journal of Magnetic Resonance, 2011. **211**(2): p. 149-155.
61. Henkelman, R.M., et al., *Quantitative interpretation of magnetization transfer*. Magnetic Resonance in Medicine, 1993. **29**(6): p. 759-66.
62. Zhang, S., et al., *PARACEST agents: modulating MRI contrast via water proton exchange*. Accounts of chemical research, 2003. **36**(10): p. 783-790.
63. Terreno, E., et al., *Highly shifted LIPOCEST agents based on the encapsulation of neutral polynuclear paramagnetic shift reagents*. Chemical Communications, 2008(5): p. 600-602.
64. Singh, A., et al., *Dependence of CEST effect from amine protons of glutamate on pH*. Proceedings of 19th ISMRM, Montreal, Canada, 2011: p. 713.
65. Chuang, K., et al., *MRI Detection of Brain Glucose Uptake using Gluco-CEST*.
66. Aime, S., et al., *Pushing the sensitivity envelope of lanthanide-based magnetic resonance imaging (MRI) contrast agents for molecular imaging applications*. Accounts of chemical research, 2009. **42**(7): p. 822-831.
67. Merbach, A.E., L. Helm, and É. Tóth, *The Chemistry of Contrast Agents in Medical Magnetic Resonance Imaging* 2013: Wiley Online Library.
68. Liu, G., et al., *Imaging In Vivo Extracellular pH with a Single Paramagnetic Chemical Exchange Saturation Transfer Magnetic Resonance Imaging Contrast Agent*. Molecular imaging, 2011.
69. Sheth, V., et al., *Measuring in vivo tumor pHe with a PARACEST MRI contrast agent*.
70. McVicar, N., et al., *Simultaneous in vivo pH and temperature mapping using a PARACEST-MRI contrast agent*. Magnetic Resonance in Medicine, 2012.
71. Lagadic-Gossmann, D., L. Huc, and V. Lecureur, *Alterations of intracellular pH homeostasis in apoptosis: origins and roles*. Cell Death and Differentiation, 2004. **11**(9): p. 953-961.
72. Martinez-Zaguilan, R., et al., *Acidic pH enhances the invasive behavior of human melanoma cells*. Clinical and Experimental Metastasis, 1996. **14**(2): p. 176-186.

73. Murakami, T., et al., *Elevated expression of vacuolar proton pump genes and cellular PH in cisplatin resistance*. International Journal of Cancer, 2001. **93**(6): p. 869-874.
74. Prescott, D.M., et al., *The relationship between intracellular and extracellular pH in spontaneous canine tumors*. Clinical Cancer Research, 2000. **6**(6): p. 2501-2505.
75. Webb, B.A., et al., *Dysregulated pH: a perfect storm for cancer progression*. Nature Reviews Cancer, 2011. **11**(9): p. 671-677.
76. Madshus, I.H., *Regulation of intracellular pH in eukaryotic cells*. Biochemical Journal, 1988. **250**(1): p. 1.
77. Delgado, R., et al., *Lanthanide complexes of macrocyclic derivatives useful for medical applications*. Pure and applied chemistry, 2005. **77**(3): p. 569-579.
78. Bryson, J.M., J.W. Reineke, and T.M. Reineke, *Macromolecular Imaging Agents Containing Lanthanides: Can Conceptual Promise Lead to Clinical Potential?* Macromolecules, 2012. **45**(22): p. 8939-8952.
79. Kiessling, F., et al., *Synthesis and characterization of HE-24.8: a polymeric contrast agent for magnetic resonance angiography*. Bioconjugate chemistry, 2006. **17**(1): p. 42-51.
80. Zhou, J., et al., *Differentiation between glioma and radiation necrosis using molecular magnetic resonance imaging of endogenous proteins and peptides*. Nature medicine, 2010. **17**(1): p. 130-134.
81. Walker-Samuel, S., et al., *Assessment of tumour glucose uptake using glucose-CEST*. 2011.
82. Desmond, K. and G. Stanisz, *pH mapping based on the ratiometric amide and amine relationship from endogenous CEST*. In Proceedings of the 3rd International Workshop on Chemical Exchange Saturation Transfer Imaging, Annapolis, MD, USA, 2012.
83. McVicar, N., et al., *Amine/Amide Concentration Independent Detection (AACID) of Intracellular pH by CEST MRI at 9.4 T*. In Proceedings of the 21st Annual Meeting of ISMRM, Salt Lake City, Utah, United States, 2013: p. 4226.
84. Lim, E.K., et al., *pH-Triggered Drug-Releasing Magnetic Nanoparticles for Cancer Therapy Guided by Molecular Imaging by MRI*. Advanced Materials, 2011. **23**(21): p. 2436-2442.
85. Bae, Y., et al., *Multifunctional polymeric micelles with folate-mediated cancer cell targeting and pH-triggered drug releasing properties for active intracellular drug delivery*. Molecular BioSystems, 2005. **1**(3): p. 242-250.
86. Ko, J., et al., *Tumoral acidic extracellular pH targeting of pH-responsive MPEG-poly ( $\beta$ -amino ester) block copolymer micelles for cancer therapy*. Journal of Controlled Release, 2007. **123**(2): p. 109-115.