# Cracked Iron Oxide Nanoprticles as T2 Contrast Agents for Magnetic Resonance Imaging

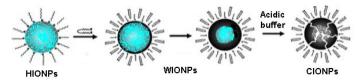
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#### **ABSTRACT**

Metal [Fe, Mn, Gd, and Co] oxide nanoparticles water suspension as magnetic resonance imaging (MRI) contrast agents has a variety of characteristics that penetrate effectively biological membranes, circulate for a long time in blood vessel, and conjugate with targeted receptors. Especially, nanoparticles with small size and large surface provide magnetic resonance image with high sensitivity and specificity at low imaging-agent concentration.[1] Metal oxide nanoparticles with hollow spheres can incorporate therapeutic agents into their payloads, enabling simultaneous MRI diagnosis and delivery of drugs to targeted sites. There have very recently been developed a novel and facile synthesis of hollow manganese oxide nanoparticles (HMONs) and their potential application as multifunctional agents for simultaneous MR imaging and drug delivery.[2] HMONs have strong relaxivities due to their large surface. Iron oxide nanoparticles as T<sub>2</sub> contrast agents are employed to image tumors, stem cell migration, and cancer metastases. Iron ions are usually safer than potentially toxic metal ions such as Gd<sup>3+</sup> and Mn<sup>2+</sup> in the body. Herein, we report a facile synthesis of nontoxic cracked iron oxide nanoparticles (CIONPs) from hydrophobic FeO nanoparticles (HIONPs) via 3 steps. With complex surface structure, CIONPs showed improved r<sub>2</sub> relaxivities compared to hydrophobic FeO nanoparticles (HIONPs). We expect that CIONPs have the potential application as a drug or chemical delivery vehicle because of their cracked spheres. In addition, cellular and in vivo MR imaging study with CIONPs will be tested.

### **EXPERIMENT**



Scheme 1. Formation of CIONPs (FeO-: sky blue, Fe<sub>3</sub>O<sub>4</sub>-: black, oleate: , poly(ethylene glycol) phospholipid,

#### Procedure for formation of 14 nm CIONPs

- (1) Synthesis of hydrophobic FeO nanoparticles (HIONPs) Thermolysis of Fe(acac)<sub>3</sub> with surfactants such as OA (oleic acid) and OAm (oleylamine)
- (2) Synthesis of hydrophilic iron oxide nanoparticles (WIONPs) PEG-phospholipid coating
- (3) Oxidation under distilled water and formation cracked Fe<sub>3</sub>O<sub>4</sub> nanoparticles (CIONPs) under acidic buffers (pH 2.6~4.6)

#### RESULTS

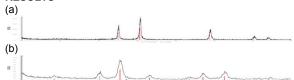


Figure 1. XRD patterns of nanoparticles. (a) Hydrophobic FeO nanoparticles spectrum (black pattern : hydrophobic Hydrophobic FeO nanoparticles, red line ; reference FeO nanoparticles. (b) Cracked Fe $_3$ O $_4$  nanoparticles spectrum (black pattern : cracked Fe $_3$ O $_4$  nanoparticles, red line : reference Fe $_3$ O $_4$  nanoparticles).

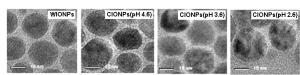


Figure 2. TEM image of WIONPs, CIONPs (pH 4.6), CIONPs (pH 3.6), and CIONPs (pH 2.6)

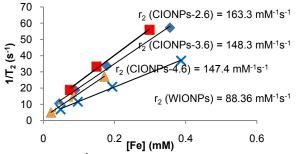


Figure 3. Plots  $T_2^{-1}$  versus Fe concentration for WIONPs (x), CIONPs-4.6 ( $\triangle$ ), CIONPs-3.6 ( $\bullet$ ), and CIONPs-2.6 ( $\blacksquare$ ).

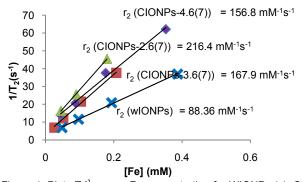


Figure 4. Plots  $T_2^{-1}$  versus Fe concentration for WIONPs (x), CIONPs-4.6 (7)( $\bullet$ ), CIONP-3.6(7) ( $\blacksquare$ ), and CIONP-2.6(7) ( $\triangle$ ).

Table 1. Relaxation of nanoparticles at 4.7 T MRI

Nanoparticles	T <sub>2</sub> <sup>[a]</sup> [ms]	$r_2[s^{-1}mM^{-1}]$
WIONPs	122	88.4
CIONPs-4.6 <sup>[b]</sup>	69	147.7
CIONPs-3.6 <sup>[b]</sup>	65	148.3
CIONPs-2.6 <sup>[b]</sup>	51	163.3
CIONPs-4.6(7) <sup>[c]</sup>	55	156.8
CIONPs-3.6(7) <sup>[c]</sup>	49	167.9
CIONPs-2.6(7) <sup>[c]</sup>	38	216.4

<sup>&</sup>lt;sup>a</sup> Measured at 0.08 mM (as measured by ICP-AES).

## CONCLUSIONS

- Cracked iron oxide nanoparticles (CIONPs) are generated form monodispersed hydrophobic FeO nanoparticles (HIONPs) via 3 steps, such as surface coating with PEG-phospholipid, oxidation under water, and getting FeO phase off using acidic buffers.
- 2. CIONPs have good T<sub>2</sub> relaxivities and potential applications, such as vectors for drug delivery and chemical storage.

### REFERENCES

- 1.S. Mornet, S. Vasseur, F. Grasset, E. Duguet, J. Mat. Chem. 2004, 14, 2161.
- 2.J. Shin, R. M. Anisur, M. K. Ko, G. H. Im, J. H. Lee, I. S. Lee, Angew. Chem. Int. Ed. 2009, 48, 321.

<sup>[</sup>b] Formatted using WIONPs under water for 1 day with acidic buffer.

<sup>[</sup>c] Formatted using WIONPs under water for 7 days with acidic buffer.