A Calcium Sensitive Magnetic Resonance Contrast Agent

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Introduction

As part of our efforts to study cell signaling and regulation in intact animals, we are developing MR contrast agents that provide information about physiological species and biochemical events.¹ Here, we report the first MRI contrast agent DOPTA-GD (Figure 1) whose relaxivity is selectively modulated by Ca^{2+} concentration.² Ca^{2+} is an important intracellular secondary messenger of signal transduction and regulates many cellular functions. A noninvasive technique for measuring Ca^{2+} changes in living organisms will serve as an important tool for biomedical research.

Results and Discussion

This new MR agent modulates access of water to a chelated Gd³⁺ ion in the presence and absence of Ca^{2+} . The design of the agent is based on the synthesis and characterization of several model systems that ultimately led to the macrocyclic dimer shown in Figure 1. 1,2-bis(o-aminophenoxy)ethane-N,N,N',N'-tetra-acetic acid (BAPTA) binds Ca^{2+} with a 10⁵ fold selectivity versus the divalent metal ion Mg^{2+} , and is relatively insensitive to pH fluctuations at physiological conditions.³ 1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane (DO3A) chelates lanthanides with high affinity to form a thermodynamically stable and kinetically inert complex.⁴ DOPTA-GD was designed to possess two limiting conformational states with respect to calcium concentration ($[Ca^{2+}]$). In the absence of Ca²⁺, each pair of aromatic iminoacetates of BAPTA interacts with Gd^{3+} through ionic attractions. In the presence of Ca²⁺, the aromatic iminoacetates of BAPTA rearrange to bind Ca^{2+} thereby allowing water to bind directly to Gd^{3+} .

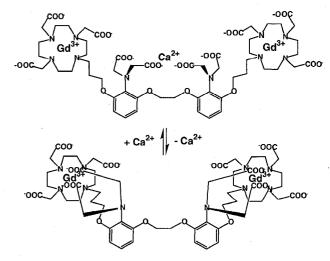
DOPTA-GD was synthesized from nitroresorcinol in 8 steps.² The effect of $[Ca^{2+}]$ on the relaxivity of DOPTA-GD was assessed by T₁ measurements. The relaxivity of DOPTA-GD in Ca²⁺ free buffer = 3.26 mM⁻¹sec⁻¹, and increased with increasing $[Ca^{2+}]$. The change in relaxivity is mostly striking in the $[Ca^{2+}]$ range of 0.1 µM to 10 µM, and levels off at higher level of $[Ca^{2+}]$ reaching a maximum of 5.76 mM⁻¹sec⁻¹. Hill plot analysis of the measured relaxivities at varying $[Ca^{2+}]$ result in a dissociation constant of the complex = 0.96 µM.

The increase in observed relaxivity of DOPTA-GD (~80%) that is induced by an increase in $[Ca^{2+}]$ corresponds to a 80% relaxivity change of each Gd³⁺ unit and is significantly higher then our previously reported enzyme-reporter class of agents.¹ In addition, the relaxivity of DOPTA-GD is relatively insensitive to

[Mg²⁺] change. Increasing Mg²⁺ concentration from 0 to 10 mM changed the relaxivity of DOPTA-GD less than 8%. Intracellular [Mg²⁺] is approximately 1 mM and its fluctuation is less dynamic than [Ca²⁺], and therefore interference on the [Ca²⁺] measurements by DOPTA-GD should be minimal. Further, changing the pH from 6.80 to 7.40 changed the measured T₁ of DOPTA-GD by less than 3% (in the presence or absence of Ca²⁺). Therefore, within physiological pH ranges, H⁺ should not interfere with the relaxivity of the complex with respect to [Ca²⁺].

In summary, we have synthesized a MR contrast agent where the relaxivity of the complex is controlled by the presence or absence of the divalent ion Ca^{2+} . By structurally modulating inner-sphere access of water to a chelated Gd^{3+} ion we observe a substantial change in T₁ upon the addition of Ca^{2+} . Importantly, the agent is selective for binding Ca^{2+} ions versus Mg^{2+} and H⁺. An immediate application of this agent is to study the cellular Ca^{2+} activity changes during the embryogenesis. The agent can be conveniently injected inside cells at the early developmental stage. Both the cell movements and the Ca^{2+} fluctuations during the experiments may help to resolve some uncertainties of measuring Ca^{2+} activity of the interior cell layers not accessible to light microscopy.

Figure 1. Schematic of DOPTA-GD representing the proposed conformational dependence on Ca^{2+} .



References: 1. Moats, R. A., Fraser, S. E., Meade, T. J. Angew. Chem. Int. Ed. Eng. 36, 726, **1997. 2.** Li, W. H., Fraser, S. E., Meade, T. J. J.Am.Chem.Soc. In press. **3.** Tsien, R. Y. Biochemistry 19, 2396, 1980. **4.** Kumar, K., Chang, C. A., Tweedle, M. F. Inorg. Chem. 32, 587,1993.